

Enhanced O₃ and NO₂ in Thunderstorm Clouds: Convection or Production?

Tanja Winterrath¹, Thomas P. Kurosu², Andreas Richter, and John P. Burrows

Institute of Environmental Physics, University of Bremen, Germany

Abstract. Ground based zenith sky measurements of O₃ and NO₂ slant optical thickness (SOT) from a field campaign at the Observatoire de Haute Provence in June 1996 are reported. Differential optical absorption spectroscopy in the 450–497 nm region was used to derive SOT for both species under clear sky, cloudy and thunderstorm conditions. SOT enhancements of 62% for O₃ and up to 320% for NO₂ are found in a thunderstorm cloud. Interpretation of the measurements was carried out using a radiative transfer model. To explain the measurements a mean increase of 38 ppbv of the in-cloud concentration of O₃ has to be assumed. This amount leads to a mean in-cloud concentration of more than 100 ppbv of O₃. The corresponding increase in NO₂ is 3 ppbv, resulting in an in-cloud concentration of 3.2 ppbv. The NO₂ enhancements are in large part explained by lightning processes. In contrast, the enhanced O₃ appears to be induced by intrusion of stratospheric air and possibly ozone production by non-lightning discharge mechanisms.

Introduction

As previous measurements have shown, the slant optical thickness (SOT) increase in the presence of optically thick clouds [Brewer and Kerr, 1973; Van Roozendaal *et al.*, 1994; Erle *et al.*, 1995]. The reason is twofold: first, the atmospheric light path is enhanced due to multiple scattering of light by the cloud particles [Kurosu *et al.*, 1997; Pfeilsticker *et al.*, 1998, and references therein]. This leads to larger SOTs, which can be calculated with an appropriate radiative transfer model such as GOMETRAN [Rozanov *et al.*, 1997]. Second, the chemical composition of the in-cloud air is different from cloud free air as a result of transport [Dickerson *et al.*, 1987], photolysis and heterogeneous processes specific to clouds [Lelieveld and Crutzen, 1990], and chemical reactions specific to thunderstorms [Ridley *et al.*, 1994; Price *et al.*, 1997].

Estimates of NO_x production resulting from lightning vary considerably, because measurements are rare and influenced by the inhomogeneity of the clouds (see Huntrieser *et al.* [1998] for a survey of previous studies). The observed mixing ratios vary between less than 2 ppbv and 60 ppbv (e.g., Drapcho *et al.* [1983], Dickerson *et al.* [1987]), all showing enhanced in-cloud mixing ratios compared to

measurements in the boundary layer. Absorption measurements show a maximum column of 5.5·10²⁰ molec/m² of NO₂ in a thunderstorm [Franzblau and Popp, 1989], and enhancements in the total column by a factor of 10–30 in NO₂ and 1.2 in O₃ during pre-monsoon thunderstorms [Jadhav *et al.*, 1996].

Measurements of O₃ in clouds indicate that both production and loss mechanisms exist. Dickerson *et al.* [1987] measured enhanced mixing ratios inside a cloud compared to the boundary layer value, Poulida *et al.* [1996] measured high O₃ near but not within a thunderstorm anvil. Both attributed this effect to intrusion of stratospheric air. Hauf *et al.* [1995] detected a drop in O₃ on entering the anvil core of a thunderstorm cloud. Similar measurements by Ridley *et al.* [1994] and Huntrieser *et al.* [1998] also showed no systematic increase of in-cloud O₃ concentrations. Sonde measurements carried out by Shlanta and Moore [1972] show O₃ values inside a cloud at 6 km that are 2.6 times higher than their pre-storm boundary layer level. They assigned the sources to be point discharges at the ground followed by transport and in-cloud production. In contrast, Drapcho *et al.* [1983] found a correlation between a decrease in O₃ and an increase in NO₂ caused by the production of NO, followed by its reaction with O₃. Recent laboratory studies of arc discharges also indicate a large ratio of NO/NO₂ and minimal O₃ production [Wang *et al.*, 1998].

The measurements presented in this paper have been performed by the Bremen group during the NDSC (Network for the Detection of Stratospheric Change) and SCUVS (Stratospheric Climate using UV/Vis-Spectroscopy) validation campaign at the Observatoire de Haute Provence (OHP), which is located at 44°N, 6°E (see Figure 1) at an altitude of 684 m above sea level. The aim of this work is to analyse the enhancement of the SOTs in O₃ and NO₂, which were measured on June 15th, 1996 at OHP, when convective cumulonimbus clouds developed over the site.

Measurements and Analysis

Detailed information about the validation campaign and the zenith sky spectrometer and DOAS (Differential Optical Absorption Spectroscopy) retrieval algorithm used for this study are given in Richter *et al.* [1995] and Roscoe *et al.* [1998]. Briefly, for each solar zenith angle (SZA) the term

$$(\ln I(\Theta_0, \lambda)/I(\Theta, \lambda)) - \sum_i \sigma_i(\lambda) SCD_i - P(\lambda)$$

is minimised in a least square fit for the selected spectral window. Here, $I(\Theta, \lambda)$ is the radiance measured at SZA Θ and wavelength λ , Θ_0 a reference SZA, σ_i the absorption cross section, $SCD_i = SOT_i/\sigma_i$ the slant column density, and $P(\lambda)$ a polynomial, which is used to differentiate broad

¹Now at Institute of Atmospheric Physics, Johannes Gutenberg University of Mainz, Germany.

²Now at Harvard-Smithsonian Center for Astrophysics, Cambridge, Massachusetts, USA.

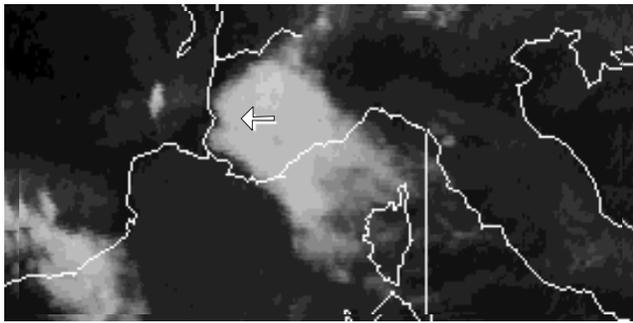


Figure 1. METEOSAT picture (provided by DWD) showing the analysed thunderstorm cloud over part of southern France at 6 pm on 06/15/96. Northern Italy is on the right, Corsica and Sardinia at the bottom. The white arrow marks the location of OHP.

band absorptions and scattering from differential gas absorption structures. The spectral window for both O₃ and NO₂ measurements was 450–497 nm. This was selected after a careful study of potential interferences by other absorbers. Figure 2 shows June 15th measurements for O₄, O₃, and NO₂, as well as the Colour Index (CI), a measure of cloudiness, defined as the ratio of the red to the blue part of a spectrum. A significant enhancement in the SOTs of all trace gases was observed during the thunderstorm. Similar results were obtained on other days, but the relative amounts of the O₄, NO₂ and O₃ enhancements varied from cloud to cloud. As part of the SCUVS project, cloud measurements of several zenith sky instruments at OHP have been compared [Pfeilsticker *et al.*, 1998]. It is important to note, that zenith sky measurements of the type described here under an optically thick cloud yield values averaged over a large volume, and cannot be readily compared to in-situ measurements.

Modelling Calculations simulating the enhancements due to changes in the radiative transfer were performed with GOMETRAN, which is a 1-dimensional full multiple scattering radiative transfer model based on the finite differencing method. It contains a quasi-exact parameterisation for cloud scattering [Kurosu *et al.*, 1997] where the cloud is defined by plane-parallel layers with an altitude dependent droplet spectrum and cloud optical properties derived from Mie theory. For the radiative transfer calculations presented here, balloon borne measurements taken by the CNRS group at Gap, close to the measurement site on the same day (temperature, pressure, O₃) and June 19th (NO₂) were used as profile input.

Determination of cloud parameters On June 15th, convection started at midday inducing the development of clouds over the site as can be seen in Figure 1. METEOSAT data clearly indicate a cumulonimbus system of large horizontal extent covering the measurement site completely at 6 pm, the time selected for detailed analysis. Lightning was observed in the afternoon starting around 4:30 pm. Heavy rain was recorded between 4 and 7 pm. The wind direction was from the south-west, turning north-east in the evening, with frequent changes of direction during the thunderstorm episode.

The cloud condensation level was estimated to be 2.2 km, assuming an adiabatic rise of air from the ground. Cloud top height was inferred from the cloud top temperature observed by METEOSAT. The cloud particle profile, which is shown in Figure 3, was derived from GOMETRAN simulations of

the SOT of the chemically inert O₄ (proportional to [O₂]²) by matching the model results for a cumulonimbus cloud to the observed SOT of O₄.

Determination of NO₂ and O₃ in-cloud mixing ratios The cloud parameters, that reproduce the observed O₄ amount, yield SOTs for NO₂ and O₃ which are significantly different from the measured quantities. Therefore the chemical composition of the cloud air mass must have changed. Table 1 shows the values for clear sky conditions and the measured and calculated amounts for the cloudy situation.

The model fit of the measurements is performed by variation of the assumed mixing ratios until the observed O₃ and NO₂ SOTs are obtained. Two cases were evaluated for this purpose: the first, where changes were limited to taking place within the cloud (case I), results in values of about 3.20 ppbv and 115 ppbv for NO₂ and O₃, respectively. For the second case, a constant mixing ratio is assumed from the ground to the top of the cloud (case II). This results in values of about 3.15 ppbv and 109 ppbv, i.e., not significantly different from case I. For NO₂, two alternative vertical profiles (IIb+c) were investigated, which also reproduced the observed SOT (see Figure 3). Profile IIb corresponds to the case when the NO/NO₂ ratio is determined by a photostationary state, involving NO₂ photolysis and the reaction of NO with O₃. In this calculation, actinic fluxes calculated by GOMETRAN for this particular cloud and sonde measured O₃ and T profiles have been used. For this case, a 38% decrease of the NO₂ mixing ratio from cloud bottom to cloud top was derived for a constant NO_x mixing ratio inside the cloud, resulting in an enhancement in the NO₂ column of $4.3 \cdot 10^{20}$ molec NO₂/m². The IIc case has a cloud bottom to cloud top ratio of 10 in NO₂, which could represent the upwelling of highly polluted air masses from the boundary layer. Even for this extreme situation, the vertical column of NO₂ is only 4% higher than in the case of the constant

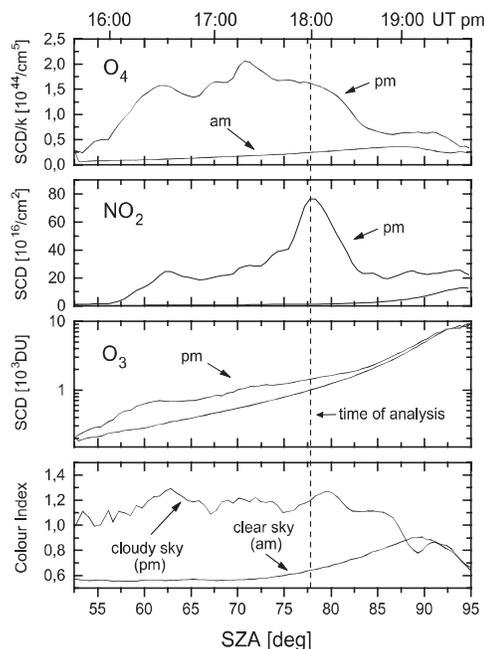


Figure 2. The time development of the observed SCDs of various trace gases and the CI (see text for definition). The evening values during the cloudy period are clearly enhanced. Note the logarithmic scale for O₃. The dashed line indicates the time of analysis.

profile. This demonstrates that the shape of the vertical profile within the cloud has only a small influence on the integrated trace gas column. In a series of sensitivity tests, the mixing ratios obtained for O₃ and NO₂ were shown to be relatively insensitive to cloud vertical position. For example, a vertical shift of ± 500 m results in changes of less than 2.0% in O₃ and less than 2.5% in NO₂.

Discussion

Three possible sources for the observed O₃ and NO₂ enhancements were studied: convective transport of polluted air masses, stratospheric intrusion, and lightning induced production.

Convection A longer time-series of measurements taken at OHP indicates that enhancements in NO₂ column are often associated with tropospheric pollution (*J.-P. Pommereau, private communication*). On June 15th, southwesterly winds transported air from the direction of Marseille towards the measurement site. To explain the enhancements observed at OHP, a Marseille plume having 3.8 ppbv of NO₂ at 2 pm is required. Measurements in the boundary layer at Cadarache, approx. 32 km S-SW of OHP have lower early afternoon values. In the late afternoon, when Cadarache was also influenced by lightning, high values of up to 10 ppbv were observed. In summary, convection may have contributed but cannot explain all of the observed increase in NO₂ column.

Assuming advection as the source of O₃ in the cloud, a boundary layer air parcel with a mixing ratio of 110 ppbv is required. An O₃ sounding, which was launched at Gap at 9 am on the 15th, shows a mixing ratio of 42 ppbv O₃ at 590 m. On the 17th, when the meteorological situation was similar to that on the 15th, a value of 63 ppbv O₃ was measured around 6 pm. Values of 110 ppbv are usually not encountered in the afternoon, as the Cadarache data

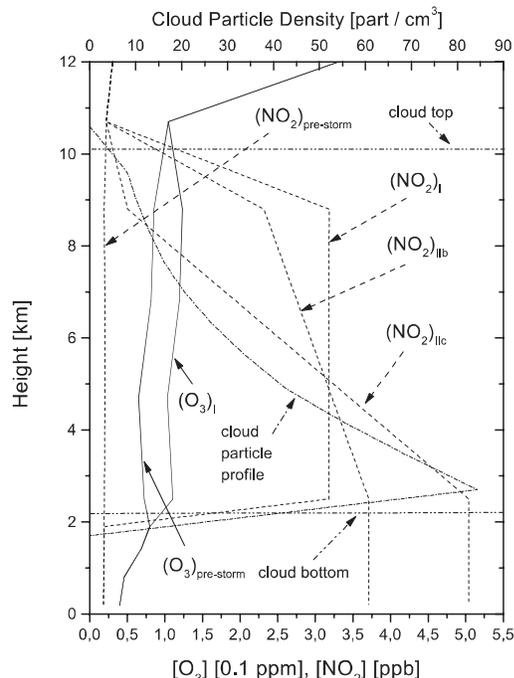


Figure 3. Trace gas profiles: pre-storm and modified to reproduce the observed cloud effect (labeled according to the different cases described in the text) and vertical profile of the cloud particle density.

Table 1. Observed (clear sky (AM) and cloudy sky (PM)) and modelled (cloudy sky) SOTs for 06/15/96. The diurnal change in stratospheric NO₂ has been taken into account by applying a factor of 1.4 to the AM NO₂ column. The calculated values are given for a cumulonimbus cloud between 2.2 and 10.1 km, and the original trace gas profiles as described in the text. All SOTs are given for 77.8° SZA and 477 nm.

SOT	clear sky	cloudy	model
O ₄	0.025	0.111	0.111
O ₃	0.023	0.037	0.032
NO ₂	0.007	0.220	0.022

confirm. Overall, it is unlikely that convection is responsible for the majority of the observed increase in O₃.

Stratospheric intrusion Transport of air parcels from the stratosphere into the cloud anvil has been identified as a source of high concentrations of O₃ measured in cloud top regions by several authors [Dickerson *et al.*, 1987; Poulida *et al.*, 1996]. and can explain part of the observed enhancement in O₃ columns. For NO₂, the same mechanism can be ruled out as a result of its small mixing ratios in the lower stratosphere.

Lightning and discharge induced production Mechanisms for the production of O₃ and NO₂ involving discharge processes have been discussed by *Griffing, 1977*. As the dissociation energy of O₂ is lower than that of N₂, molecular nitrogen only dissociates as a result of lightning flashes, whereas oxygen may also be dissociated by non-lightning discharge processes. Although lightning itself does not appear to produce O₃ [Wang *et al.*, 1998], processes such as coronal, or silent discharges are potential sources [Bhethanabhotla *et al.*, 1985]. O₃ production is therefore likely to occur prior to lightning in an electrically active cloud. Comparing the time development of the SOTs, two observations can be made: first, the maxima in O₃ and NO₂ do not coincide. O₃ peaks before NO₂, in agreement with the measurements of *Jadhav et al.* [1996]. Second, at approximately 78° SZA, a strong enhancement in NO₂ occurs, that is not observed in O₃. Convection of polluted air masses is an unlikely explanation of this effect.

Thunderstorm properties and the magnitude of lightning induced NO₂ production have been calculated according to the method of *Price et al.* [1997], using the measured temperature profile and our derived cloud properties as input data. As a result of the large uncertainty obtained when applying statistical data to a particular cloud, these calculations only yield order of magnitude estimates. For our conditions, this results in a statistical frequency of 2.1 flashes per minute, a NO_x production rate of $6.7 \cdot 10^{26}$ molec/cloud-to-ground lightning (CG) and a proportion of CG of 42%. The NO_x production for CG and intra-cloud lightning is calculated to be $6.8 \cdot 10^{26}$ molec/min. This results in a total production of $6.1 \cdot 10^{28}$ molec NO_x or a column of $6.9 \cdot 10^{20}$ molec NO_x/m² at 6 pm. Assuming a ratio of 0.86 for the vertical columns of NO₂/NO_x, as calculated assuming photostationary state (case IIb), a value of $5.9 \cdot 10^{20}$ molec NO₂/m² results. The enhancement in the vertical column of NO₂ from the ground to the cloud top, as derived from GOMETRAN calculations, is $4.3 \cdot 10^{20}$ molec NO₂/m². The two values are

of the same order of magnitude, but, as values of the NO_x production rate found in the literature cover a wide range [Biazar and McNider, 1995], this agreement might nevertheless be fortuitous. In conclusion, a significant contribution to the overall NO₂ content in the cloud appears to result from lightning discharge.

According to Griffing [1977], 3·10²⁶ molec O₃/flash are produced. This value results in an enhancement of the vertical column of O₃ between ground and cloud top of 6.2·10²⁰ molec O₃/m². According to Jadhav et al. [1996], lightning yields no more than 2–3% of the total O₃ produced in a thunderstorm cloud. This implies that a total O₃ column for the OHP cloud would be 2–3·10²² molec/m². The measured excess O₃ vertical column at 6 pm on the 15th of June 1996 was 5.1·10²¹ molec O₃/m². Both the temporal behaviour and this estimate indicate that O₃ might be produced by non-lightning discharges in the cloud. However, intrusion of stratospheric air could equally well explain the observed enhancement.

Summary

Zenith sky measurements of enhanced NO₂ and O₃ SOTs in a thunderstorm cloud have been analysed using the radiative transfer model GOMETRAN. Mixing ratios in the thunderstorm of more than 100 ppbv O₃ and 3 ppbv NO₂ were derived. Neglecting changes in the in-cloud chemistry, three mechanisms are likely to account for these enhancements: convective transport from the boundary layer, stratospheric intrusion, and discharge induced production. The differences in the temporal behaviour of O₃ and NO₂ indicate that the observed enhancements of the two molecules have different origins. For NO₂, lightning is the most likely source for the majority of the enhanced NO₂ observed at 6 pm. A smaller contribution may in addition result from the convection of polluted air. For O₃, intrusion of stratospheric air and possibly O₃ production by non-lightning discharge processes appear to be the most likely explanation of the observed in-cloud amount and its temporal behaviour. This study has used ground based DOAS measurements to detect in-cloud trace gas composition. It appears a promising approach for the study of lightning and related processes in clouds.

Acknowledgments. The authors would like to thank the Deutscher Wetterdienst for providing METEOSAT pictures; the organisers and participants of the SCUVS / NDSC intercomparison campaign at OHP, especially F. Goutail and J.-P. Pommereau, who also provided SAOZ measurements, local trace gas data and meteorological information, C. Vialle for O₃ sondes data, and H. K. Roscoe; the SCUVS cloud sub group lead by K. Pfeilsticker; and D. W. Arlander for additional weather information. Helpful comments from R. R. Dickerson, K. Pickering and J.-P. Pommereau are gratefully acknowledged. Parts of this work have been funded by the University of Bremen, the German Ministry of Education and Research (BMBF), and the European Union.

References

- Bhetanabhotla, M. N., et al., Simulation of trace species production by lightning and corona discharge in moist air, *Atmos. Environ.*, 19(9), 1391–1397, 1985.
- Biazar, A. P., and R. T. McNider, Regional estimates of lightning production of nitrogen oxides, *J. Geophys. Res.*, 100(D11), 22861–22874, 1995.
- Brewer, A. W., and J. B. Kerr, Total ozone measurements in cloudy weather, *Pure Appl. Geophys.*, 106, 928–937, 1973.
- Dickerson, R. R., et al., Thunderstorms: An important mechanism in the transport of air pollutants, *Science*, 235, 460–464, 1987.
- Drapcho, D. L., et al., Nitrogen fixation by lightning activity in a thunderstorm, *Atmos. Environ.*, 17(4), 729–734, 1983.
- Erle, F., et al., On the influence of tropospheric clouds on zenith-scattered-light measurements of stratospheric species, *Geophys. Res. Lett.*, 22, 2725–2728, 1995.
- Franzblau, E., and C. J. Popp, Nitrogen oxides produced from lightning, *J. Geophys. Res.*, 94(D8), 11089–11104, 1989.
- Griffing, G. W., Ozone and oxides of nitrogen production during thunderstorms, *J. Geophys. Res.*, 82(6), 943–950, 1977.
- Hauf, T., et al., Rapid vertical trace gas transport by an isolated midlatitude thunderstorm, *J. Geophys. Res.*, 100(D11), 22957–22970, 1995.
- Huntrieser, H., et al., Transport and production of NO_x in electrified thunderstorms: Survey of previous studies and new observations at midlatitudes, *J. Geophys. Res.*, 103(D21), 28247–28264, 1998.
- Jadhav, D. B., et al., Observations of NO₂ and O₃ during thunderstorm activity using visible spectroscopy, *Adv. Atmos. Sci.*, 13(3), 359–374, 1996.
- Kurosu, T., et al., Parameterization schemes for terrestrial water clouds in the radiative transfer model GOMETRAN, *J. Geophys. Res.*, 102(D18), 21809–21823, 1997.
- Lelieveld, J., and P. J. Crutzen, Influences of cloud photochemical processes on tropospheric ozone, *Nature*, 343, 227–233, 1990.
- Pfeilsticker, K., et al., Optical path modifications due to tropospheric clouds: Implications for zenith sky measurements of stratospheric gases, *J. Geophys. Res.*, 103(D19), 25323–25335, 1998.
- Pfeilsticker, K., et al., Intercomparison of the detected influence of tropospheric clouds on UV-visible absorptions detected during the NDSC intercomparison campaign at OHP in June 1996, *this issue*
- Poulida, O., et al., Stratosphere-troposphere exchange in a midlatitude mesoscale convective complex 1. Observations, *J. Geophys. Res.*, 101(D3), 6823–6836, 1996.
- Price, C., et al., NO_x from lightning, 1. Global distribution based on lightning physics, *J. Geophys. Res.*(D5), 102(D5), 5929–5941, 1997.
- Richter, A., et al., Ground based UV/vis measurements of O₃, NO₂, BrO, and OClO over Bremen (53°N), in *Proceedings of the Third European Workshop on Stratospheric Ozone, Schliersee 1995* 1995.
- Ridley, B. A., et al., Distribution of NO, NO_x, NO_y, and O₃ at 12 km altitude during the summer monsoon season over New Mexico, *J. Geophys. Res.*, 99(D12), 25519–25534, 1994.
- Roscoe, H. K., et al., Slant column measurements of O₃ and NO₂ during the NDSC intercomparison of zenith sky UV/Vis spectrometers in June 1996, *J. Atmos. Chem.*, in press, 1998.
- Rozanov, V., et al., GOMETRAN: A radiative transfer model for the satellite project GOME - the plane-parallel version, *J. Geophys. Res.*, 102(D14), 16683–16695, 1997.
- Shlanta, A., and C. B. Moore, Ozone and point discharge measurements under thunderclouds, *J. Geophys. Res.*, 77(24), 4500–4510, 1972.
- Van Roozendaal, M., et al., Ground-based visible measurements at the Jungfraujoch station since 1990, *J. Quant. Spec. and Radiat. Transf.*, 52, 231–240, 1994.
- Wang, Y. et al., Nitric oxide production by simulated lightning: Dependence on current, energy, and pressure, *J. Geophys. Res.*, 103(D15), 19149–19159, 1998.

T. Winterrath, Institute of Atmospheric Physics, Johannes Gutenberg University, Becherweg 21, D-55099 Mainz, Germany (e-mail: winterra@mail.uni-mainz.de)

T. P. Kurosu, Harvard-Smithsonian Center for Astrophysics, 60 Garden Street, Cambridge, MA 02138, USA

A. Richter, J. P. Burrows, Institute of Environmental Physics, University of Bremen, PO Box 33 04 40, D-28334 Bremen, Germany

(Received October 26, 1998; revised December 28, 1998; accepted January 21, 1999.)